

<b>Study Programme:</b> Chemistry			
<b>Course Unit Title:</b> Organic Chemistry IV			
<b>Course Unit Code:</b> IHO-301			
<b>Name of Lecturer(s):</b> Assistant professor Aleksandar Oklješa			
<b>Type and Level of Studies:</b> Bachelor Academic Studies			
<b>Course Status (compulsory/elective):</b> Elective			
<b>Semester (winter/summer):</b> Winter			
<b>Language of instruction:</b> English			
<b>Mode of course unit delivery (face-to-face/distance learning):</b> Face-to-face			
<b>Number of ECTS Allocated:</b> 5			
<b>Prerequisites:</b> None			
<b>Learning objectives</b> Obtaining knowledge of Molecular Orbital (MO) theory and the principle of hard and soft acids and bases (HSAB) as the most accessible approaches to understanding many aspects of reactivity. Gaining knowledge and skills in planning of organic reactions and their applications in modern organic synthesis.			
<b>Learning outcomes</b> Qualifying students to create and interpret organic reactions for the synthesis of the planned compounds.			
<b>Syllabus</b> <i>Theoretical instruction</i> Molecular Orbital (MO) theory and structures of organic molecules. The Principle of Hard and Soft Acids and Bases (HSAB). Factors affecting the position of an equilibrium and chemical reactivity. Ionic Reactions – Reactivity. Pericyclic Reactions (sigmatropic rearrangement, Diels-Alder reactions, [3+2]- and [2+2] cycloaddition reactions. The Woodward-Hoffmann Rules. Photochemical Reactions. <i>Practical instruction</i> Laboratory synthesis of organic compounds.			
<b>Required Reading:</b> 1. G. Proctor, Stereoselectivity in organic synthesis, Oxford, University press, 1998			
<b>Weekly Contact Hours:</b> 60	<b>Lectures:</b> 30	<b>Practical work:</b> 30	
<b>Teaching Methods:</b> Lectures and laboratory work			
<b>Knowledge Assessment (maximum of 100 points):</b> 100			
<b>Pre-exam obligations</b>	points	<b>Final exam</b>	points
Activity	5	Written exam	35
Lab exercises	25	Oral exam	15
Seminar work	20		