

Course Unit Descriptor

Study Programme: Bachelor Academic Studies in Biochemistry			
Course Unit Title: Medicinal chemistry			
Course Unit Code: B-404			
Name of Lecturer(s): Professor Velimir Popsavin			
Type and Level of Studies: Bachelor of Science Degree			
Course Status (compulsory/elective): compulsory			
Semester (winter/summer): summer			
Language of instruction: English			
Mode of course unit delivery (face-to-face/distance learning): Face-to-face			
Number of ECTS Allocated: 9			
Prerequisites: None			
Course Aims: Acquiring the basic methods and biochemical principles relevant to the development, processing and biological testing of new pharmacologically active molecules – potential drugs.			
Learning Outcomes: Students will be able to identify structural features essential for the pharmacological activity of potential drugs and to understand their biological effects on the molecular level.			
Syllabus: <i>Theory</i> Methods and objectives of medical chemistry. A brief overview of ligand-receptor interactions that are important for pharmacological effects of drugs at the molecular level. The general stages in drug discovery and design. Leads and analogues: some desirable properties. Sources of leads and drugs. Methods and routes of administration: the pharmaceutical phase. Introduction to drug action. Classification of drugs. Prodrugs. Stereochemistry and drug design. Solubility and drug design: solubility and the structure of the solute; salt formation; the incorporation of water solubilising groups in a structure. Partition. Structure-activity relationship (SAR): changing size and shape; introduction of new substituents; changing the existing substituents of lead. Quantitative structure-activity relationship (QSAR). Introduction to computer-aided drug design. <i>Practice</i> In accordance with theoretical instruction.			
Required Reading: 1. V. Popsavin: Fundamentals of Medicinal chemistry, internal script (ePMF), 2019. 2. G. L. Patrick: <i>An Introduction to Medicinal Chemistry</i> , 4 th Edition, Oxford University Press, Oxford, 2009. 3. G. Thomas: <i>Medicinal Chemistry – An Introduction</i> , 2 nd Edition, John Wiley & Sons, Ltd, Chichester, 2007. 4. T. Nogrady: <i>Medicinal Chemistry: A Biochemical Approach</i> , 2 nd Edition, Oxford University Press, Oxford, 1988.			
Weekly Contact Hours: 105	Lectures: 45+15	Practical work: 45	
Teaching Methods: Lectures, laboratory work, seminar(s)			
Knowledge Assessment (maximum of 100 points): 100			
Pre-exam obligations	points	Final exam	points

Active class participation	10	written exam	70
Practical work	10	oral exam	
Preliminary exam(s)		
Seminar(s)	10		
The methods of knowledge assessment may differ; the table presents only some of the options: written exam, oral exam, project presentation, seminars, etc.			