Study program/study programs: **Bachelor of Science in Biochemistry**, **Bachelor of Science in Environmental Protection**

Type and level of studies: Bachelor

Course name: Organic chemistry I

2nd semester

Teacher: Assistant professor Srđan Bjedov

Course status: **Obligatory**

Number of ECTS credits: 8

Requirement: None

Course aim

Course Organic chemistry 1 provides a systematic study of the theories, principles, and techniques of organic chemistry. Topics include nomenclature, structure, properties, reactions, and mechanisms of the important classes of organic molecules. It is designed, when followed by Organic Chemistry 2, to fulfill the organic chemistry requisites for chemistry and biochemistry majors.

Course code: Z103

The major aim of this course is to introduce students to the foundations of organic chemistry by focusing on the structures, properties, and chemical reactivity of the various classes of organic molecules. This course will also cover mechanisms of organic reactions, and different aspects of isomerism and introduce various analytical techniques (MS, NMR, and IR) used to determine organic structures.

Course outcome

On completion of the course, the student should be able to:

1. analyze the structure of organic compounds by recognizing main functional groups, naming the compounds using the I.U.P.A.C. system, and predicting their properties using the type of bonding, hybridization state, intermolecular forces, and stereochemistry;

2. describe mechanisms of reactions: free radical, nucleophilic substitution, elimination, and electrophilic addition, and apply this knowledge to predict the major product in organic reactions;

3. analyze the nature of a reagent: as a nucleophile, free radical, or electrophile, and use this knowledge to propose the synthesis of organic compounds; and

4. demonstrate proficiency in organic laboratory skills as they pertain to chemical information, safe handling, use, and disposal of organic compounds; synthetic procedures, including isolation, purification, and structure elucidation of obtained products; stoichiometry and use of instrumentation; and writing of laboratory notebooks and reports by current scientific journals styles.

Course content

- 1. Introduction to Organic Chemistry
- 2. Chemical Bonding: A Review
- 3. Alkanes and Cycloalkanes: Structure, Nomenclature, Properties, Sources, and Conformation
- 4. Alcohols and Alkyl Halides: Structures, Nomenclature, Properties, Preparation, and Mechanisms of Reactions
- 5. Alkenes: Structure, Nomenclature, Properties, Preparation, and Mechanisms of Reactions
- 6. Stereochemistry: Definitions, Analysis, and Reactions of Stereogenic Compounds
- 7. Nucleophilic Substitution Reactions
- 8. Alkynes: Structure, Nomenclature, Properties, Preparation, and Reactions
- 9. Dienes: Structure, Nomenclature, and Reactions. Polymers and Polymerization
- 10. Arenes and Aromaticity: Structure, Nomenclature, and Reactions
- 11. Carbonyl compounds: Structure, Nomenclature, and Reactions
- 12. Carboxylic acids and derivatives: Structure, Nomenclature, Properties, Preparation, and Reactions
- 13. Amines and Heterocyclic compounds: Structure, Nomenclature, Properties, and Reactions
- 14. Biomolecules: Structure, Nomenclature, Properties, and Reactions

15.	Analytical techniques in organic chemistry: Mass spectrometry, Nuclear magnetic resonance spectroscopy,
and Infra	ared spectroscopy

Literature

- 1. K. P. C. Vollhardt, N. E. Schore: Organic chemistry 6th Edition
- 2. D. Klein: Organic chemistry 3rd Edition.

Number of classes of active teaching

Lectures: 4 (105) Practice: 3 (60)

Teaching methods

Lectures, laboratory work, desk study projects, seminar

Assessment of knowledge (maximum of 100 points): 100

Other classes

Pre-exam obligations			Points	Final exam	Points		
activity during lecture classes			5	written exam	60		
practical teaching			15				
midterm exams			2×10				
Attendance policy:							
Lectures and Laboratory Practice are mandatory.							
Grading scale:							
Grade	Points	Descri	Description				
10	91–100	Outsta	Outstanding participation in the course; excellent comprehension				
		and pre	and presentation of the course content;				
9	81–90	Very g	Very good involvement in the course; good working knowledge				
		of the learning outcomes;					
8	71-80	Good participation in the course; reasonable understanding of					
		the course content;					
7	61–70	Satisfactory engagement in the course; working knowledge of the course material;					
6	51-60	Partial	Partial engagement in the course; performance meet minimum criteria;				
5	Under 50	Inadeq	Inadequate involvement in the course; fractional understanding				
		and kn	and knowledge of the learning outcomes;				